

Download Chemical Names And Formulas Review

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CHAPTER 9 Chemical Names And Formulas Test Review ...

CHAPTER 9 Chemical Names And Formulas Test Review. 1. when the name of the anion (X) ends in -ide the acid name begins with the prefix hydro-. The stem of the anion has the suffix -ic and is followed by the word acid (ex. HCl= hydrochloric acid) 2. when the anion name ends in -ite the acid name is the stem of the anion with...

Grade 10 Chemical Names and Formulas Review – Teacher ...

Grade 10 Chemical Names and Formulas Review. Include: ionic bonds and covalent bonds S2-2-03 Write formulas and names of binary ionic compounds. Include: IUPAC guidelines and rationale for their use. S2-2-04 Write Formulas and names of molecular compounds using prefixes. Include: mono, di, tri and tetra.

Review Sheet: Chemical Names and Formulas

In writing a formula for an ionic compound, the net ionic charge of the formula must be zero. Name the following compounds and tell what type of compound they are (binary ionic, ternary ionic, or binary molecular).

Chemical Names and Formulas

Printable Chemistry Worksheets - Chemical Names and Formulas. In order to complete these worksheets, you may wish to review the rules for naming ionic compounds and covalent compounds. Other printable chemistry worksheets are also available.

SparkNotes: SAT Chemistry: Chemical Formulas Review ...

Naming Simple Compounds. Monatomic, metal, cation: simply the name of the metal from which it is derived. Al 3+ is the aluminum ion (these are often referred to as group A metals). Transition metals form more than one ion ; Roman numerals (in parentheses) follow the ion's name. Cu 2+ is copper (II) ion.

Chemistry Assignments Chapter 5: Chemical Names and Formulas

Review/practice test. Thurs Feb 6 A. Fri Feb 7 B Test names/formulas. Assignments Chapter 9: Chemical Names and Formulas Mrs. Hammond. Date Lesson Homework Thurs Jan 16 A. Fri Jan 17 B. Formulas, types of particles, binary ionic compounds. p 256 # 1-2 p 258 # 8-9 . p 263 # 10-11 .

ANSWERS TO FINAL REVIEW: CHEMICAL NAMES AND FORMULAS

ANSWERS TO FINAL REVIEW: CHEMICAL NAMES AND FORMULAS 1. Name these compounds: LiCl lithium chloride K₂S potassium sulfide AlI₃ aluminum iodide ZnS zinc sulfide BaF₂ barium fluoride Mg₃N₂ magnesium nitride MgO magnesium oxide Na₂O sodium oxide H₃P hydrogen phosphide Au₃P gold (I) phosphide NH₄F ammonium fluoride K.

Chapter 9 Practice Test

OBJ: 9.3.1 Interpret the prefixes in the names of molecular compounds in terms of their chemical formulas. | 9.3.2 Apply the rules for naming and writing formulas for binary molecular compounds. STA: Ch.2.a

Name Date Class CHEMICAL NAMES AND FORMULAS 9

CHEMICAL NAMES AND FORMULAS. 9. The ionic charge is numerically equal to the group number. The charge of an ion in Groups 5A, 6A, or 7A is determined by subtracting 8 from the group number. Stock system lesser greater The Stock system gives the actual charge of the ion.